Investigations on the formation of positive and negative reactant ions in High Kinetic Energy Ion Mobility Spectrometry (HiKE-IMS) by HiKE-IMS-MS

Maria Allers*1, Ansgar T. Kirk1, Duygu Erdogdu2, Robin Hillen2, Walter Wissdorf2, Thorsten Benter2, Stefan Zimmermann1

¹ Leibniz University Hannover, Institute of Electrical Engineering and Measurement Technology, Appelstr. 9A, 30167 Hannover, Germany, *phone: +49 511 762 4589, *e-mail: allers@geml.uni-hannover.de ² University of Wuppertal, Department of Physical and Theoretical Chemistry, Gauss Str. 20, 42119 Wuppertal, Germany

Introduction

Similar to PTR-MS or SIFDT-MS, HiKE-IMS ionize in a reaction region operated at decreased pressures around 20 mbar and high reduced electric field strengths of up to 120 Td. However, instead of a mass spectrometer, an IMS operated at the same pressure as the reaction region is used for ion separation and detection. In this work, the prevailing reactant ion population in HiKE-IMS is analyzed in dependence on the reduced electric field strength in the reaction region E_{RR}/N and the sample gas humidity. To identify individual ion species associated with certain peaks in the ion mobility spectrum, the HiKE-IMS is coupled to a time-of-flight mass spectrometer (TOF-MS) using a simple gated interface allowing to transfer selected peaks of the ion mobility spectrum into the TOF-MS.

For detailed information please see references [1], [2] and [3].

Experimental setup Drift region Quadrupole ion guide Time-of-Flight mass spectrometer Reaction region Corona gate 2 14.2 mbar 14.2 mbar 2·10-2 mbar 6·10-6 mbar E_{DR}/N Membrane pump Turbo pump Turbo pump **1** Analyte gas HiKE-IMS

- » Operating pressure: 14.2 mbar; operating temperature: 45 °C.
- » Positive ion polarity: $E_{DR}/N = const. = 115$ Td, $E_{RR}/N = variable$; Negative ion polarity: $E_{DR}/N = const. = 80$ Td, $E_{RR}/N = variable$.
- » Unless otherwise stated, drift and sample gas: Purified air containing 70 ppm, of water (0.35 % rH) and 30 ppm, of carbon dioxide.

Conclusion

The experimental data suggest the predominant generation of the positive reactant ions NO+, H_3O^+ and O_2^+ as well as the negative reactant ions O^- , OH^- , O_2^- , and O_3^- in purified air containing 70 ppm, of water and 30 ppm, of carbon dioxide. Both the reduced electric reaction field strength and the sample gas humidity significantly affect their relative abundances.

References & Acknowledgment

- [1] Allers, Kirk, von Roßbitzky, Erdogdu, Hillen, Wissdorf, Benter, Zimmermann, "Analyzing Positive Reactant Ions in High Kinetic Energy Ion Mobility Spectrometry (HiKE-IMS) by HiKE-IMS-MS", J. Am. Soc. Mass Spectrom., 2020, 31, 812-821.
- [2] Allers, Kirk, Eckermann, Schaefer, Erdogdu, Wissdorf, Benter, Zimmermann, "Positive Reactant Ion Formation in High Kinetic Energy Ion Mobility Spectrometry (HiKE-IMS)", J. Am. Soc. Mass Spectrom., 2020, Just accepted.
- [3] Allers, Kirk, Timke, Erdogdu, Wissdorf, Benter, Zimmermann, "Negative Reactant Ion Formation in High Kinetic Energy Ion Mobility Spectrometry (HiKE-IMS)", J. Am. Soc. Mass Spectrom., 2020, Under review.

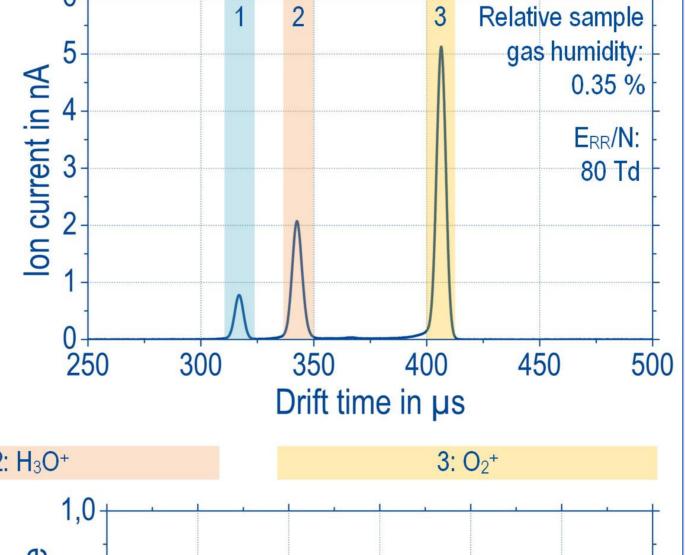


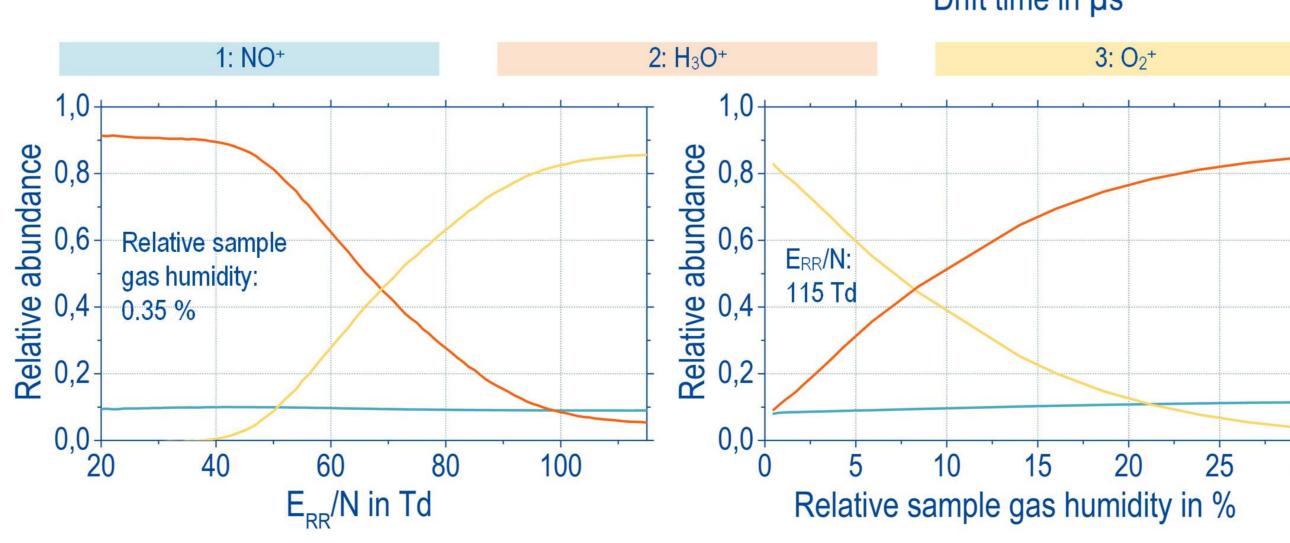
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Positive reactant ions

- positive reactant ion population is determined by the conversion processes of O_2^+ and NO^+ to $H_3O^+(H_2O)_n$.
- » Increasing E_{RR}/N inhibits the conversion, increasing sample gas humidity favors it.





Negative reactant ions

- » The negative reactant ion population is determined by the _-0,5 interaction of electrons with = -1,0neutrals in the reaction region. ত্রি-1,5-
- » Increasing the sample gas humidity results in generation of $OH^{-}(H_2O)_n$ ions.

1: O- or OH-

